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PHYSICAL CHEMISTRY Letters



Effect of Structural Phase Transition on Charge-Carrier Lifetimes and Defects in CH₃NH₃Snl₃ Perovskite

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Supporting Information

ABSTRACT: Methylammonium tin triiodide (MASnI₃) has been successfully employed in leadfree perovskite solar cells, but overall power-conversion efficiencies are still significantly lower than for lead-based perovskites. Here we present photoluminescence (PL) spectra and timeresolved PL from 8 to 295 K and find a marked improvement in carrier lifetime and a substantial reduction in PL line width below ~110 K, indicating that the cause of the hindered performance is activated at the orthorhombic to tetragonal phase transition. Our measurements therefore suggest that targeted structural change may be capable of tailoring the relative energy level alignment of defects (e.g., tin vacancies) to reduce the background dopant density and improve charge extraction. In addition, we observe for the first time an above-gap emission feature that may arise from higher-lying interband transitions, raising the prospect of excess energy harvesting.



rganic-metal-halide perovskites have emerged as a new class of solution-processable direct-bandgap semiconductors for photovoltaic device applications.¹⁻⁵ Solar cells with lead-based perovskites have now reached remarkable power conversion efficiencies^{6,7} (PCEs) of over 20%, owing to long charge-carrier diffusion lengths^{8,9} and benign defect chem-istry;^{10,11} however, there have been some environmental concerns about using these materials in large-scale applications because of the possibility of lead contamination due to leakage from damaged solar panels.^{12–15} As a lead-free alternative, tinbased perovskites have been implemented in solar cells.¹⁶⁻²¹ For methylammonium tin triiodide (MASnI₃) an open-circuit voltage of 0.88 V has been reported, which is high compared with the band gap of around 1.2 to 1.3 eV;¹⁶ however, the short-circuit currents and fill factors remain low with maximum reported values of 21 mA cm⁻²¹⁸ and 0.48,¹⁷ respectively, and the record PCE stands at 6%,¹⁶ far below that for lead-based perovskite photovoltaics. It has been suggested that the low performance of tin-based perovskites is caused by the low formation energy of tin vacancies,^{20,22,23} resulting in excessive p-doping. Several studies have shown that both MASnI₃ and CsSnI₃ can become spontaneously hole-doped during crystal formation with dark carrier densities ranging from 10^{14} to 10^{19} cm⁻³ depending on preparation method.^{17,18,21,22,24,25} Such doping may lead to strong monomolecular charge-carrier decay components^{16,21} from electron recombination with the large background hole density, which may limit the carrier diffusion and extraction.^{16,19,21,24} First-principles calculations for CsSnI₃ have suggested that tin vacancies, and indeed all acceptor-type defects, have energy levels close to the valence band edge and are therefore easily ionized, thus p-doping the material.

Despite the importance of such tin vacancies, experimental studies of their nature and their effect on the optoelectronic properties are relatively scarce. Here we have investigated such links by monitoring the photoluminescence (PL) and absorption spectra and the PL lifetime of MASnI₃ as a function of temperature from 8 to 295 K. We find that at room temperature the charge lifetime is short, and the emission line width is severely broadened in accordance with the presence of a sizable density of defects that act as hole donors. As the temperature is reduced below the phase transition into the orthorhombic phase near 110 K, a significant line narrowing and recovery of charge-carrier lifetimes is observed, and an additional emission peak originating from higher energy transitions appears. These abrupt changes coincide with a previously reported structural phase transition²⁴ and a decrease in hole concentration,²⁶ which indicates that crystal structure, hole doping, disorder, and carrier lifetime are intrinsically linked. Such improved understanding of links between defect chemistry and crystal structure may therefore highlight routes toward further optimization of tin-based perovskites.

We deposited $MASnI_3$ perovskite on z-cut quartz disks by spin-coating using methylammonium iodide and tin(II) acetate precursor solutions (see Supporting Information). Because of the known instability of tin perovskites in air, we prepared the samples and mounted them into a cryostat inside a nitrogenfilled glovebox. We carried out measurements under vacuum, with samples for PL measurements never being exposed to air and samples for absorption measurements being exposed to air for less than a minute during transfer. Measurements were taken starting from room temperature (295 K) and cooling with liquid helium to 8 K. We observed no change to the original room-temperature properties once the sample was

Received:February 12, 2016Accepted:March 18, 2016Published:March 18, 2016

returned to room temperature after cooling, showing that no permanent damage had occurred.

We first examine the general shape of the absorption and emission spectra of $MASnI_3$ as a function of temperature. Figure 1a shows a map of the PL intensity as a function of



Figure 1. (a) Color map of photoluminescence intensity as a function of temperature, normalized to the peak intensity at each temperature. We excited the sample with a pulsed laser at 400 nm with a fluence of 3 nJ cm⁻² and collected spectra at intervals of 5 K, cooling the sample from room temperature. (b) Photoluminescence spectra on a semilog scale and (c) absorption spectra (without reflection correction) from high temperature (red) to low temperature (blue) in the range 8–295 K. Dashed vertical lines mark the two peak PL energies at 8 K.

temperature between 8 and 295 K. The peak emission at room temperature is found to occur at 1.3 eV, near the range of theoretical calculations²⁷ and previous measurements^{16–18,25} of the band gap energy. As the temperature is decreased, the PL peak shifts to lower energy, similar to previously reported band gap trends in $CsSnI_3^{28}$ and $MAPbI_3^{29}$ This gradual red shift is mirrored in the absorption spectrum (Figure 1c) obtained from measuring the optical transmission of $MASnI_3$ using Fourier transform infrared (FTIR) spectroscopy. We observe a clear absorption onset corresponding to the transition at the band gap (T1), which is centered ~200 meV above the PL peak emission energy, suggesting a sizable Stokes shift.

MASnI₃ is known to exhibit phase transitions at 275^{30,31} and 108 K³¹ when cooled (but 114 K when heated), transitioning from pseudocubic (α) at room temperature²⁵ to tetragonal (β) to orthorhombic (γ) at low temperature. Interestingly, we observe no discontinuity in absorption or PL energy with temperature at the phase transitions, which suggests that the band-edge states do not substantially change here. This behavior vastly differs from that at the β to γ phase transition (160 K) in MAPbI₃, which results in a 100 meV shift in bandgap energy;^{29,32} however, as we show later, the phase transition in MASnI₃ at ~110 K is predominantly associated with changes in impurity-induced disorder and charge-carrier recombination dynamics.

Before we analyze the effect of disorder and impurities on the optoelectronic properties of $MASnI_3$ we first highlight a peculiar occurrence, which is the appearance of a secondary, higher-energy absorption and emission feature near 1.75 eV, marked as T2 in Figure 1b. This second PL peak is generally very weak (note the logarithmic scale); it emerges at 110 K and grows in intensity with respect to the main peak as the temperature is decreased further, reaching an intensity of one hundredth of that of the main peak at the lowest temperature. In contrast, a matching absorption feature for T2 is seen over the entire temperature range (Figure 1c), which red shifts in the same way as the band edge (Figure 2a). Therefore, this



Figure 2. (a) Energies of the two transitions apparent in the absorption and emission spectra of MASnI₃, T1 (absorption, squares; emission circles) and T2 (absorption, diamonds). Emission energies were taken from PL peaks, and absorption energies were taken from the midpoint of the broadened absorption rise, as determined by finding the maximum gradient. Note that the midpoint is not exactly equal to the band gap but gives a direct indication of how the band gap shifts with temperature. (b) Line width of PL peak T1, determined by fitting a Gaussian to the PL spectra (see Figure S2). The dashed black line shows the equation $\Gamma(T) = \gamma_{imp} e^{-E_b/kT} + \Gamma_0$ with $E_b = 10$ meV, $\gamma_{imp} = 119$ meV, and $\Gamma_0 = 103$ meV.

higher-energy transition T2 appears to be intrinsic to the band structure of $MASnI_3$, persisting over the entire temperature range but only yielding efficient emission in the orthorhombic phase at low temperature.

We note that a similar secondary onset is also seen in the optical absorption spectrum of $MAPbI_3$ (at 2.6 eV or 480 nm)

and found to coincide with a photobleach signal in transient absorption spectra,^{33–35} but luminescence at this wavelength has never been reported. This transition has been variously attributed to an additional valence band,³³ to a different valley in the reciprocal lattice,³⁶ or to a charge-transfer state.³⁴ A further photobleach signal in MAPbI₃ is sometimes seen at ~2.43 eV (510 nm) and attributed to excess PbI₂;³⁷ however, SnI₂ has a band gap of 2.57 eV (at 8 K),³⁸ far from the peak we observe at 1.75 eV; therefore, T2 cannot be caused by excess SnI₂.

We therefore attribute the origin of the emission observed at around 1.7 to 1.8 eV to additional transitions associated with the electronic band structure of MASnI₃. Here the most likely candidate is a transition between the highest-lying valence band maximum (VB1) and the minimum of the split-off conduction band (CB2) at the zone center (see Figure S5). Theoretical calculations predict that spin orbit coupling is important in the band structure of both lead and tin perovskites, resulting in two conduction bands, CB1 and CB2, split by the energy of the spin-orbit interaction E_{SOC} .³⁹ For MASnI₃, $E_{SOC} = 0.4$ eV, compared with 1.5 eV for MAPbI₃.⁴⁰ Two optically allowed transitions are therefore associated with these bands-a strong transition from VB1 to CB1, associated with band gap energy E_g = 1.3 eV, and another weak transition from VB1 to CB2 with energy $E = E_g + E_{SOC} = 1.7$ eV.^{39,40} These theoretical predictions correspond to the energies of the transitions we observe, making this a likely explanation. Alternatively, these transitions could arise from different electronic bands, or different points in the Brillouin zone (Figure S5); however, the latter would require exceptionally slow intraband carrier thermalization, which is unlikely. We note that the proposed mechanism of emission from charge-carrier recombination involving split-off bands has also been observed in the direct semiconductor GaAs,^{41,42} which highlights the similarity of these perovskites with fully inorganic semiconducting materials.

We continue to examine the optoelectronic properties of $MASnI_3$ by investigating changes in the emission line width with temperature, which reveals the influence of disorder and impurities. The PL line width (Γ) was determined from the full width at half-maximum (FWHM) of the emission spectrum and is shown in Figure 2b as a function of temperature. We find Γ for MASnI₃ to be rather large, with a value of 180 meV at room temperature compared with 103 meV for MAPbI₃⁴³ and 50 meV for CsSnI₃.⁴⁴ In general, both homogeneous line width broadening (typically from phonons) and inhomogeneous broadening (from impurities and energetic disorder) may contribute to the overall width of the emission spectrum. The resulting emission line width is hence described by the following equation⁴⁵

$$\Gamma(T) = \gamma_{\rm ac}T + \gamma_{\rm LO}/(e^{E_{\rm LO}/kT} - 1) + \gamma_{\rm imp}e^{-E_{\rm b}/kT} + \Gamma_0$$

Here the terms (in order of appearance) arise from scattering with acoustic phonons (coupling strength γ_{ac}), optical phonons of energy $E_{\rm LO}$ (coupling $\gamma_{\rm LO}$), donor/acceptor impurities with ionization energy $E_{\rm b}$ (coupling $\gamma_{\rm imp}$), and a temperature-independent broadening term Γ_0 arising, for example, from energetic disorder. In the high-temperature range the phonon terms increase linearly with T (see Figure S6 for example curves), which is contrary to the sublinear dependence of the line width on temperature we observe. Such sublinear behavior is, however, accurately described by the functional form of the ionizable impurity term in the above equation, suggesting that

scattering of charge-carriers from ionized impurities such as tin vacancies is the predominant contribution to line width broadening in MASnI₃ above 110 K. Because the impurity terms dominate, we neglect phonon contributions and fit the line width from 140 to 270 K (within the tetragonal phase), which gives a binding energy of $\sim 10 \text{ meV}$ (Figure 2), although the small-temperature range means that binding energies from 10 to 20 meV give reasonable fits (Figure S7); however, this functional form alone cannot account for the surprisingly abrupt line width narrowing below 110 K. We propose that because the line width in this material is dominated by impurities, the sudden change must be associated with a sharp decrease in number or change in binding energy of the defects below the phase transition. As previously mentioned, the most easily formed point defects in tin iodide perovskites are tin vacancies.^{22,23} Removing a tin atom can contribute two holes to the valence band²² if the resulting defect energy level lies close to the valence band, as predicted.²³ At sufficiently low temperature $(kT < E_{\rm b})$, however, holes can potentially become localized to their acceptor ions (such as tin vacancies), making the defect charge-neutral and lowering the interaction with charge-carriers, as described in the above model. Hence the emission line width is expected to narrow as dopant charge carriers localize. The abrupt change at the phase transition may indicate that a structural change makes it favorable for the holes to become localized, resulting in a sudden decrease in the density of holes in the valence band. We note that this explanation agrees well with the increase in electrical resistivity that has previously been observed^{24,26} when MASnI₃ was lowered below the transition into the orthorhombic phase at 110 K. The large change in line width at the phase transition, despite the absence of strong changes in band gap energy, is surprising; however, it suggests that relatively small changes in structure can suffice to suppress excess dopant carrier density in tin halide perovskites.

The temperature-independent inhomogeneous line width value ($\Gamma_0 = 103 \text{ meV}$) approached at very low temperature reflects the sizable disorder remaining in this phase, despite the improvement at the 110 K transition. The large contribution to the emission broadening in MASnI₃, from both ionizable defects (γ_{imp}) and temperature-independent disorder (Γ_0) , is in stark contrast with MAPbI₃ for which the broadening is thought to be almost entirely homogeneous at room temperature.⁴³ Similar conclusions can be made when comparing the Stokes shifts between the absorption onset and the emission peak observed for these two materials. Here we find a large Stokes shift of almost 200 meV for MASnI₃ independent of temperature (Figure 2a), whereas for MAPbI₃, Stokes shifts of at most 10 meV have been reported,²⁹ in accordance with significantly higher energetic disorder in the tin perovskite. Such contrast in defect densities has also been reported between copper indium gallium sulfide (CIGS) and copper zinc tin sulfide (CZTS), where defects in CZTS result in tail states that protrude into the band gap.^{46,47} Tail states cause the peak PL energy to be considerably red-shifted from the absorption onset because carriers tend to thermalize into the tail states, which is typically accompanied by an undesirable lowering of charge-carrier mobility.

Finally, we investigated the effect of temperature-dependent doping and disorder on the charge-carrier lifetime. We measured time-resolved PL transients using time-correlated single photon counting (TCSPC) after excitation with laser pulses of 400 nm wavelength, as shown in Figure 3a. Because



Figure 3. (a) Time-resolved photoluminescence (PL) as a function of time after excitation, from high temperature (red) to low temperature (blue). The instrument response function is shown by the shaded gray area. (b) Top: PL lifetime determined as the time to reach 1/e of the maximum intensity. Bottom: relative intensity of the second transition (T2) intensity in comparison with that originating from the band gap (T1).

the PL lifetime is very short, it is important to take account of the instrument response function (IRF) shown by the gray shaded area, which was obtained by measuring the response of the detector to laser pulses at 990 nm (near the detection wavelength). Because a single functional form for the decay, convoluted with the IRF, could not fit the entire temperature range satisfactorily we chose to use the 1/e lifetime to quantitatively compare the decays. We smoothed the data and IRF and deconvoluted the two using a fast Fourier transform algorithm to obtain a corrected decay curve (see Figure S1). Using this method, we determined the time τ_{e} taken for the PL to decay to 1/e of its initial value, shown in Figure 3b as a function of temperature. The PL lifetime changes very little at temperatures above 110 K, remaining fixed at ~45 ps, which is considerably shorter than lifetimes for MAPbI₃ (typically on the order of 100-1000 ns).⁴⁸ The short carrier lifetime at room temperature is problematic for planar device architectures, which require a long diffusion length for carrier extraction, although if the carrier mobilities are high enough, planar architectures will still be achievable.^{16,49} In addition, we find the shape of the room-temperature PL transients to be independent of excitation fluence over a large measured range $(0.3-300 \text{ nJ cm}^{-2})$, as shown in Figure S4. This behavior implies that monomolecular recombination is heavily dominant, rather than bimolecular electron-hole recombination that would result in fluence-dependent dynamics.¹¹ Monomolecular recombination is expected to dominate when there is a high level of defects that act as recombination centers¹¹ or there is a

high background carrier concentration such that the photoexcited carrier concentration has little influence on the dynamics.¹⁶ Hence the observed fast carrier recombination dynamics in this temperature region is compatible with the broad emission line width observed and the presence of a large tin vacancy density.

Below \sim 110 K the lifetime increases steeply, reaching a value of ~ 275 ps at 8 K. Furthermore, the dynamics become dependent on excitation fluence here (see Figure S4) for the same investigated range, suggesting the background carrier concentration is now lower. This explanation correlates well with the observed steep decline in emission line width previously discussed. If, as we have already suggested, there is a change in the energy levels of the acceptor ions such that they are elevated further above the valence band maximum, it will become suddenly more favorable for the holes to be localized to these energy levels as the temperature is decreased, causing the hole density to drop. It is worth noting that a rise in lifetime at low temperature is also seen in CIGS and CZTS, with CZTS showing a thousand-fold increase.⁴⁷ This is attributed to spatial electrostatic variations due to high levels of disorder or defects, which confines the carriers in different locations, providing a barrier to recombination that is overcome thermally as the temperature is increased. Hence charge-carrier lifetimes can sensitively depend on both recombination with a density of impurity-induced background carriers and the energetic disorder associated with such impurities.

Interestingly, this rise in carrier lifetime below the phasetransition temperature concurs with the observed increase in intensity of the secondary PL peak T2 (Figure 3b), alluding to a common cause. This could simply be due to the interplay between the rate constants for recombination at the transitions T1 and T2. While T1 may be mostly dominated by electron recombination with the sizable dark hole density, T2 could be largely influenced by phonon-mediated interband transitions, for example, between multiple conduction (or valence) bands. As a result, the relative rates may therefore be affected differently at the phase transition, imparting the observed changes in relative emission intensity (Figure 3b).

Taken together, these observations point to a significant change between the high- and low-temperature regimes, which is induced by the phase transition at 108 K. We suggest that all of these features, the increase in PL lifetime, the drop in FWHM, the appearance of PL from a secondary transition, and the previously observed decrease in hole concentration,²⁶ are related to the presence of defects in the system (likely tin vacancies that have a low formation energy²²). These defects scatter carriers, giving the PL a broad line width, and also pdope the material, resulting in a large background hole density, which, in turn, leads to rapid charge-carrier recombination. Our observations suggest that upon the transition into the orthorhombic phase, the relative alignment of the defects' energy level must change subtly with respect to the valence band edge. This change in energy level results in background holes becoming localized to defects that reduce their contribution to carrier scattering (emission line width decreases) and also reduce the p-doping (carrier lifetime increases). Further reduction in temperature freezes out more background holes such that the lifetime increases further. An understanding of the precise mechanisms inducing these effects is important because it may hold the clue to achieving similar effects at room temperature. Scattering between charge-carriers and impurities leads to reduced charge-carrier mobilities, while

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charge-recombination with background holes reduces chargecarrier lifetimes. In combination, these effects account for the significantly lowered charge-carrier diffusion length of only a few tens of nanometers observed for MASnI₃ at room temperature.¹⁶ Our measurements indicate that while the absorption onset energy hardly changes at the phase transition near 110 K, the relative energetic alignment of defect levels with respect to the valence band edge is significantly altered. An understanding of how this favorable alignment is induced by the structural changes could potentially allow for such effects to be replicated at room temperature through compositional changes.

In summary, we have evaluated the optoelectronic properties of MASnI₃ as a function of temperature to reveal the influence of impurities on charge-carrier scattering and recombination lifetimes. We find an abrupt improvement as the temperature is lowered into the orthorhombic phase below 108 K, where emission line widths narrow and lifetimes increase significantly. We suggest that this effect is caused by a change to the energy level alignment of the tin vacancies with respect to the valence band maximum, leading to a reduction in background dopant density. First-principle calculations, that relate these effects to the structural changes anticipated at the phase transition, could hold the key to further material improvement. Because the perovskite structure allows for a range of organic/inorganic cations and halide anions to be combined with Sn²⁺, a material combination may be found that replicates the favorable change observed at the phase transition at room temperature, allowing for low background hole densities. In addition, we have observed a weak but distinct above-gap luminescence peak that coincides with the energy of a secondary absorption onset that is seen in similar form in many other hybrid metal halide perovskites. We propose that this emission originates from charge-carrier recombination involving a higher-lying band, for example, split off by spin-orbit coupling, similar to what has been observed for GaAs. The observation of such emission suggests that a certain fraction of charges exists for an appreciable time in a higher-energy state. If such excess energy could be harnessed, this could offer a further efficiency boost and provide a means for exceeding the Shockley-Queisser limit

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.jpclett.6b00322.

Experimental methods, data fitting and analysis, fluencedependent PL decay curves, schematics of recombination, and sample emission broadening curves. (PDF)

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Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

We acknowledge the Engineering and Physical Sciences Research Council (EPSRC) for financial support. E.S.P thanks the EPSRC Centre for Doctoral Training in New and Sustainable Photovoltaics (CDT-PV) for support through a doctoral studentship. T.S. acknowledges financial support by a Marie Curie Intra European Fellowship (MESO-SUPER-CELLS, PIEG-GA-2012-327015) within the 7th European Community Framework Programme.

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